

Aim :- To find out 1 cation and 1 anion in the given salt number 1.

Physical characteristics

- State :- solid
- Colour :- colourless
- Solubility :- soluble in water

S.No.	Experiment	Observation	Inference
1.	Action of heat :- A pinch of given salt was taken in dry test tube and heated.	white sublimate was formed.	$\text{NH}_4^+$ may be present.
2.	Charcoal cavity test :- The given salt was mixed with double amount of sodium carbonate. Mixture was placed at charcoal drop of $\text{H}_2\text{O}$ is added to it. Mixture was heated to reducing of burner.	White mass was obtained.	$\text{Pb}^{2+}$ , $\text{Xn}^{2+}$ , $\text{Cu}^{2+}$ , $\text{Al}^{3+}$ absent.
3.	Cobalt Nitrate Test :- 1-2 drops of cobalt nitrate was added to white mass of charcoal cavity and heated in oxidising flame.	colourless white mass was obtained.	$\text{Al}^{3+}$ , $\text{Zn}^{2+}$ , $\text{Mg}^{2+}$ , $\text{PO}_4^{3-}$ absent.
4.	Flame Test :- The given salt was mixed with conc. $\text{HCl}$ in watch glass to make a paste and the paste was heated on pl wire in the	No characteristics colour was obtained.	$\text{Ca}^{2+}$ , $\text{Ba}^{2+}$ , $\text{Sr}^{2+}$ absent.

Teacher's Signature \_\_\_\_\_

# Golden Guide Of Chemistry For Class Xith

**Yan Bai**



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## **Table of Contents Golden Guide Of Chemistry For Class Xith**

1. Understanding the eBook Golden Guide Of Chemistry For Class Xith
  - The Rise of Digital Reading Golden Guide Of Chemistry For Class Xith
  - Advantages of eBooks Over Traditional Books
2. Identifying Golden Guide Of Chemistry For Class Xith
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Golden Guide Of Chemistry For Class Xith
  - User-Friendly Interface
4. Exploring eBook Recommendations from Golden Guide Of Chemistry For Class Xith
  - Personalized Recommendations
  - Golden Guide Of Chemistry For Class Xith User Reviews and Ratings
  - Golden Guide Of Chemistry For Class Xith and Bestseller Lists

5. Accessing Golden Guide Of Chemistry For Class Xith Free and Paid eBooks
  - Golden Guide Of Chemistry For Class Xith Public Domain eBooks
  - Golden Guide Of Chemistry For Class Xith eBook Subscription Services
  - Golden Guide Of Chemistry For Class Xith Budget-Friendly Options
6. Navigating Golden Guide Of Chemistry For Class Xith eBook Formats
  - ePub, PDF, MOBI, and More
  - Golden Guide Of Chemistry For Class Xith Compatibility with Devices
  - Golden Guide Of Chemistry For Class Xith Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Golden Guide Of Chemistry For Class Xith
  - Highlighting and Note-Taking Golden Guide Of Chemistry For Class Xith
  - Interactive Elements Golden Guide Of Chemistry For Class Xith
8. Staying Engaged with Golden Guide Of Chemistry For Class Xith
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Golden Guide Of Chemistry For Class Xith
9. Balancing eBooks and Physical Books Golden Guide Of Chemistry For Class Xith
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Golden Guide Of Chemistry For Class Xith
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Golden Guide Of Chemistry For Class Xith
  - Setting Reading Goals Golden Guide Of Chemistry For Class Xith
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Golden Guide Of Chemistry For Class Xith
  - Fact-Checking eBook Content of Golden Guide Of Chemistry For Class Xith
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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