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fifth
edition



Christina Latham-Koenig Clive Oxenden
Jerry Lambert with Jane Hudson

English File Intermediate Key

**Clive Oxenden, Christina Latham-
Koenig, Paul Seligson, Tracy Byrne**

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English File Clive Oxenden,Christina Latham Koenig,Gill Hamilton,1999

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Chomacki, Jane Hudson, 2025-10-16 English File 3e Intermediate Plus Workbook with Key Christina Latham-Koenig, Jane Hudson, Clive Oxenden, 2014-02 *English File: Intermediate. Student's Book with Online Practice* Christina Latham-Koenig, Clive Oxenden, Kate Chomacki, 2018-12-13 **English File 4E Intermediate Work Book without answers** Christina Latham-Koenig, Clive Oxenden, Kate Chomacki, Jerry Lambert, 2020-07-28 English File s unique lively and enjoyable lessons are renowned for getting students talking In fact 90% of English File teachers we surveyed in our impact study found that the course improves students speaking skills **New English File** Clive Oxenden, Christina Latham-Koenig, Paul Seligson, Tracy Byrne, 2006 **Language Course Planning** Brian North, Mila Angelova, 2018-04-30 This essential guide examines course planning as an end to end process from learners needs through to assessment taking into account both the broader issues and the practical details at every stage Areas covered include effective needs analysis using the CEFR as a resource for course planning writing scenarios for classroom teaching and assessment triangulating course objectives materials and learners goals key terminology Extra resources are available on the website www.oup.com/elt/teacher/lcp Brian North is a co author of the CEFR and of its companion volume and was Chair of Eequals from 2005 to 2010 Mila Angelova is the Academic Vice Chair of Eequals and Head Director of Studies at AVO Language and Examination Centre in Sofia Elzbieta Jarosz is a member of the Eequals Certification Panel and is the Academic Director of Gama College in Krakow Richard Rossner is a co founder of Eequals and a co author of the European Profiling Grid and the Eequals Framework

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web $m \times c \times \Delta t$ equation 1 where m is the total mass of the solution solute plus solvent c is the specific heat of the solution and Δt is the observed temperature change the specific heat of the solution is generally assumed to be the same as that of
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web the heat of reaction ΔH_{soln} is written after the products in units of kJ mol^{-1} rxn for example sodium hydroxide dissolves exothermically releasing 44.2 kilojoules per mole dissolved this process is represented as $\text{NaOH (s)} \rightarrow \text{NaOH (aq)}$ $\Delta H_{\text{soln}} = 44.2 \text{ kJ mol}^{-1}$ rxn write equations to similarly represent the dissolving process for

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web background the hand warmers are filled with sodium acetate click the metal disc and you will notice the liquid changes state and becomes a solid in the process heat is released the sodium acetate is a supersaturated solution

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web the goal of my experiment and research is to evaluate the various number of compounds to be used as hand warmers and locate the ones that can raise the temperature to 20 degrees celsius in 100 g water and costs as little as possible

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web in this lab you are challenged to use chemistry to determine which ionic compound makes the best hand warmer in this experiment you will collect data that will allow you to calculate the heat of solution ΔH_{soln} or the amount of energy released or absorbed in kJ mol of the substance being dissolved

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