

## Naming Compounds Practice Problems

### Naming Complex (polyatomic) Ionic Compounds

Name the following compounds:

- 1)  $\text{NH}_4\text{Cl}$  ammonium chloride
- 2)  $\text{Fe}(\text{NO}_3)_3$  iron (III) nitrate
- 3)  $\text{Pb}(\text{SO}_4)_2$  lead (IV) sulfate
- 4)  $\text{Ag}_3\text{PO}_4$  silver phosphate
- 5)  $\text{Be}(\text{HCO}_3)_2$  beryllium hydrogen carbonate
- 6)  $\text{Al}(\text{CN})_3$  aluminum cyanide
- 7)  $\text{Mn}_2(\text{SO}_3)_3$  manganese (III) sulfite
- 8)  $\text{Sr}(\text{C}_2\text{H}_3\text{O}_2)_2$  strontium acetate
- 9)  $\text{Ti}(\text{CN})_4$  titanium (IV) cyanide
- 10)  $\text{YClO}_3$  yttrium chlorate

Write the formulas for the following compounds:

- 1) lead (IV) sulfate  $\text{Pb}(\text{SO}_4)_2$
- 2) silver cyanide  $\text{AgCN}$
- 3) copper (II) chlorate  $\text{Cu}(\text{ClO}_3)_2$
- 4) chromium (IV) phosphate  $\text{Cr}_2(\text{PO}_4)_3$
- 5) vanadium (IV) carbonate  $\text{V}(\text{CO}_3)_2$
- 6) ammonium oxide  $(\text{NH}_4)_2\text{O}$
- 7) tin (II) nitrite  $\text{Sn}(\text{NO}_2)_2$
- 8) chromium (III) hydroxide  $\text{Cr}(\text{OH})_3$
- 9) titanium (II) acetate  $\text{Ti}(\text{C}_2\text{H}_3\text{O}_2)_2$
- 10) cobalt (III) chromate  $\text{Co}_2(\text{CrO}_4)_3$

# Naming Compounds Practice With Answers

**Didier Musso**



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