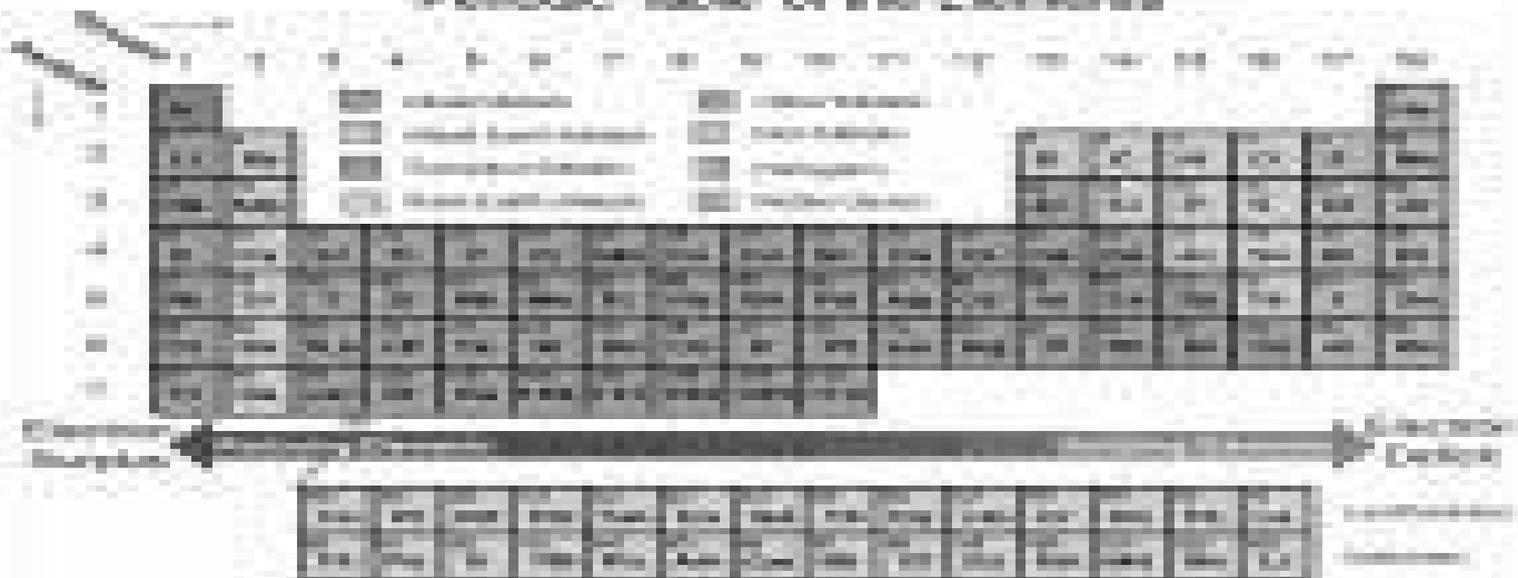


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Periodic Table of the Elements



- Period** - is the horizontal row of elements in the periodic table.
ex - Chlorine has period 3 and atomic number 17, while Fluorine has period 2.
- Group** - is the vertical row of elements in the periodic table.
ex - Fluorine has period 2 and atomic number 9, while Chlorine has period 3.
- Electronegativity** - atom's ability that you see increase as you move right and up from left to right.
ex - Fluorine has the highest electronegativity and Chlorine has the next highest.
- Ionization Energy** - The ability or tendency of an atom to lose an electron.
ex - Fluorine has the highest ionization energy and Chlorine has the next highest.

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