



Student Exploration: Balancing Equations

(Any answers are indicated within.)

Essential Question: How do we balance chemical equations? What are the coefficients in a balanced equation? How do we balance a chemical equation? What are the coefficients in a balanced equation?

What You Will Do: You will balance chemical equations and determine the coefficients.

1. **Balance** the chemical equation for the reaction of hydrogen gas and oxygen gas to form water. How many molecules of hydrogen gas and oxygen gas are needed to form 2 molecules of water? How many molecules of hydrogen gas and oxygen gas are needed to form 4 molecules of water? How many molecules of hydrogen gas and oxygen gas are needed to form 6 molecules of water? How many molecules of hydrogen gas and oxygen gas are needed to form 8 molecules of water? How many molecules of hydrogen gas and oxygen gas are needed to form 10 molecules of water?
2. **Balance** the chemical equation for the reaction of hydrogen gas and oxygen gas to form water. How many molecules of hydrogen gas and oxygen gas are needed to form 2 molecules of water? How many molecules of hydrogen gas and oxygen gas are needed to form 4 molecules of water? How many molecules of hydrogen gas and oxygen gas are needed to form 6 molecules of water? How many molecules of hydrogen gas and oxygen gas are needed to form 8 molecules of water? How many molecules of hydrogen gas and oxygen gas are needed to form 10 molecules of water?

What You Will Do:

Use the following information to balance the chemical equation for the reaction of hydrogen gas and oxygen gas to form water. How many molecules of hydrogen gas and oxygen gas are needed to form 2 molecules of water? How many molecules of hydrogen gas and oxygen gas are needed to form 4 molecules of water? How many molecules of hydrogen gas and oxygen gas are needed to form 6 molecules of water? How many molecules of hydrogen gas and oxygen gas are needed to form 8 molecules of water? How many molecules of hydrogen gas and oxygen gas are needed to form 10 molecules of water?

Chemical Equation: $2H_2 + O_2 \rightarrow 2H_2O$

1. **Balance** the chemical equation for the reaction of hydrogen gas and oxygen gas to form water. How many molecules of hydrogen gas and oxygen gas are needed to form 2 molecules of water? How many molecules of hydrogen gas and oxygen gas are needed to form 4 molecules of water? How many molecules of hydrogen gas and oxygen gas are needed to form 6 molecules of water? How many molecules of hydrogen gas and oxygen gas are needed to form 8 molecules of water? How many molecules of hydrogen gas and oxygen gas are needed to form 10 molecules of water?

Hydrogen	Oxygen	Water	Hydrogen	Oxygen
2	1	2	2	1
4	2	4	4	2
6	3	6	6	3
8	4	8	8	4
10	5	10	10	5

Essential Question: How do we balance chemical equations? What are the coefficients in a balanced equation? How do we balance a chemical equation? What are the coefficients in a balanced equation?



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A Loxley



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