

HYPERION USER GUIDE

Last updated: 02/29/2016

1. LOGGING IN

- a. Open **Firefox**  (Remote XUL Manager add-on must be installed see appendix A).
 - Hyperion is not compatible with Apple computers or with Google Chrome.
 - Hyperion can be used with **Explorer 10** or older version.
- b. Type in the following link: <https://budgeting.utrgv.edu/workspace/index.jsp>
- c. Type in your UTRGV identifier and password.
 - The UTRGV identifier is an 6 digit name with 3 letters and 3 numbers



2. At the home screen, click on the UTRGV link under the Quick Links section.



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Yicheng Fang



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Atmospheric Correction of Hyperspectral Data Wesley Jeremiah Moses, 2005 Manual of Astronomy Robert William Shaw, Samuel Latimer Boothroyd, 1902 *The Home Manual. Everybody's Guide in Social, Domestic, and Business Life. A Treasury of Useful Information for the Million ...* Mrs. John A. Logan, 1889

Image Processing and Pattern Recognition in Remote Sensing S. G. Ungar, Shiyi Mao, Yoshifumi Yasuoka, 2003 *Microcomputing*, 1984 *The Bookman's Manual* Bessie Graham, 1928

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web among the lanthanides in addition to 3 states 2 states are shown by nd sm eu tm and yb only whereas 4 state is exhibited by ce pr nd tb and dy elements rest five elements show only 3 states oxidation states of lanthanide elements it has been shown that the lanthanide elements are highly electropositive and form essentially ionic

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web five of the elements la ce pr nd eu are very reactive and when exposed to air react with oxygen to form an oxide coating that tarnishes the surface for this reason these metals are stored under mineral oil the remainder of the lanthanides are not as reactive and some gd lu retain their silvery metallic appearance for a long time

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web jun 30 2023 elements lanthanum 57 through lutetium 71 on the periodic table are considered to be lanthanides lanthanides tend to react with oxygen to form oxides the reaction at room temperature can be slow while heat can cause the reaction to

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web to separate the lanthanides from other elements occurring with them they are chemically combined with specific substances to form lanthanide compounds with low solubility oxalates and fluorides for example a process known as ion exchange is then used to separate the lanthanides from each other in this process a solution of the

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web jan 1 1997 the lanthanides comprise the largest naturally occurring group in the periodic table there are more than 100 minerals known to contain lanthanides but the only two of commercial importance are monazite and bastnaesite the chapter discusses the production of lanthanide elements by electrolysis of fused salts and by metallothermic reduction

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web the fifteen elements shown in table pageindex 1 from lanthanum la 4f 0 to lutetium lu 4f 14 are lanthanoids ln may be used as a general symbol for the lanthanoid elements although lanthanoids scandium sc and yttrium y are sometimes called rare earth elements they are relatively abundant in the earth s crust

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