

Q.57	<p>Two different enzyme catalysis reactions I and II have identical Y-intercepts for the Lineweaver-Burke (equation given below) plots. The slope for reaction I is twice than that of reaction II.</p> <p>If the initial concentrations of enzymes in I and II are same, the correct statement(s) is(are)</p>
	$\frac{1}{v} = \frac{1}{v_{max}} + \frac{K_M}{v_{max}} \frac{1}{[S]}$ <p>where v and v_{max} are rate and maximum rate; K_M is Michaelis-Menten constant, and $[S]$ is substrate concentration.</p>
(A)	Reactions I and II have same turn over number
(B)	Michaelis-Menten constants for reactions I and II are identical
(C)	Michaelis-Menten constant for reaction I is twice than that of reaction II
(D)	The rates of the elementary steps for reactions I and II are identical

Q.58	<p>The enthalpy change for the exothermic reaction between BeI_2 and HgF_2 is _____ kJ mol^{-1} (rounded off to the nearest integer)</p> <p>(Given: Bond dissociation energy (in kJ mol^{-1}) for $\text{Be-F} = 632$, $\text{Be-I} = 289$, $\text{Hg-F} = 268$ and $\text{Hg-I} = 145$)</p>

Gate Chemistry Question Paper

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