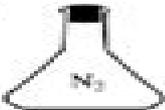
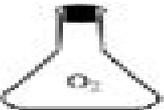


Unit 2 Gases Study Guide

Name: _____

Multiple Choice

Questions 1-3 refer to the 1 L flasks shown below. Provide an explanation for your answer.

(A)		(B)		(C)		(D)		(E)	
	T = 30°C P = 1 atm		T = 50°C P = 0.5 atm		T = 40°C P = 2 atm		T = 50°C P = 1 atm		T = 40°C P = 0.5 atm

- In which flask do the molecules have the greatest average speed?
(A) A
(B) B
(C) C
(D) D
(E) E
- Which flask contains the smallest amount of moles?
(A) A
(B) B
(C) C
(D) D
(E) E
- Which flask contains the sample with the greatest density?
(A) A
(B) B
(C) C
(D) D
(E) E
- Under which of the following conditions of temperature and pressure will H_2 gas be expected to behave most like an ideal gas?
(A) 50 K and 0.10 atm
(B) 50 K and 5.0 atm
(C) 500 K and 0.10 atm

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