

12 • The Gas Laws**THE COMBINED GAS LAW**

In practical terms, it is often difficult to hold any of the variables constant. When there is a change in pressure, volume and temperature, the combined gas law is used.

$$\frac{P_1 V_1}{T_1} = \frac{P_2 V_2}{T_2} \quad \text{or} \quad P_1 \times V_1 \times T_2 = P_2 \times V_2 \times T_1$$

$$K = ^\circ\text{C} + 273$$

*See
scratch
paper*

Complete the following chart.

	P_1	V_1	T_1	P_2	V_2	T_2
1	1.50 atm	3.00 L	20.0°C 293 K	2.50 atm		30.0°C 303 K
2	720. torr	256. mL	25.0°C 298 K		250. mL	50.0°C 323 K
3	600. mmHg	2.50 L	22.0°C 325 K	760. mmHg	1.80 L	
4		750. mL	0.00°C 273 K	2.00 atm	500. mL	25.0°C 298 K
5	95.0 kPa	4.00 L		101. kPa	6.00 L	498.°C 471. K or
6	650. torr		100.°C 373 K	900. torr	225. mL	150.°C 423 K
7	850. mmHg	1.50 L	15.0°C 288 K		2.50 L	30.0°C 303 K
8	125. kPa	125. mL		100. kPa	100 mL	75.0°C 348 K

Gas Law Worksheet Answer Key

Camilla Rothe



Gas Law Worksheet Answer Key:

Gas Law Worksheet Answer Key Book Review: Unveiling the Magic of Language

In a digital era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Gas Law Worksheet Answer Key**," written by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound affect our existence. Throughout this critique, we will delve to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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