

Elements and Compounds

Mixtures vs. Compounds

1. Mixtures:

- Definition: Made up of two or more substances that are physically combined, and their components can be separated by physical means.
- Ex. Saltwater: Saltwater is a mixture composed of salt (sodium chloride) dissolved in water. The salt and water can be physically separated by processes such as evaporation, leaving behind the salt.

2. Compounds:

- Definition: Chemically combined substances that cannot be separated by physical means.
- Ex. Sodium Chloride (NaCl): Sodium chloride, commonly known as table salt, is a compound formed by the ionic bond between sodium (Na) and chlorine (Cl) atoms. It exists as a crystalline solid and cannot be separated into sodium and chlorine by physical means.

Elements Compounds And Mixtures Unit Study Guide

Kenneth W. Watkins



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