

My Career Goals Essay Example

5 pages | 1085 words

I tried to be circumspect in my deeds since was child. Maybe, I took after my father this feature because he is a businessman and should be proactive in making decisions. I wasn't that type of a child who didn't want to go to school and study, I liked doing homework, studying, and even writing various essays. When I was 14, I made my serious choice. I decided to get an academic education and become a businessman in the future. Since that time I tried to be specific in my goals and plans trying to get the best of possible opportunities.

My Career Goals During University

Many students perceive university life as a sort of fun. And it is no wonder because it is the time when you meet new people who become your soul mates forever. It is when you have serious relationships, break-ups, quarrels, even make a family. I understood that university life is valuable because we learn the basics of how to effectively communicate with people, work in teams, groups, solve problems, work with time management, make realistic plans. All this help you achieve success in an adult life. Once I attended a lecture lead by the CEO of one the top well known local private equity fund. He told that he received his education in Columbia business school. Unfortunately, that education wasn't enough to launch his own business. This man said that our university goals should be in learning more than what university can ever give us. It was something that amazed me. His words made me look at my long term goals even in a more perseverant way. I understood that in such a volatile modern world, the university education was not enough. I tried never to miss a chance to enroll in extra curricular activities. For instance, I attended open lectures in local business schools, additional lessons in management teams, sales team, Saturday discussions about the global economy, psychology courses, a center for entrepreneurship. I tried to absorb all information connected with the business world, market, innovations because as a future businessman I had to be well-rounded. It was the most significant part of my university goals in term of getting a theoretical education. I didn't stop only on studying the material because business is connected with practice. When I was a senior student at the University of Economics in Prague at the Faculty of Business Administration, I became more involved in family business. My father owns a company which does in-depth research on innovations, cooperates and searches competitive start-ups. When I was free from studying my father explained to me how his business works, gave me professional advice. He offered me a part-time job as a junior business development manager. I worked on weekends learning about market research, networking with other companies, negotiating with potential clients. By doing this, I wanted to discover target companies which could be useful in my future private business. I wanted to continue my father's business. My target towards career was to expand his company to other countries, and this will help reach my personal career goal. I've...

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Robert R. Redfield



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for the nitration 4.01 ml of nitric acid 4.00 ml of sulfuric acid and 4.56 ml bromobenzene were used to complete the reaction with these measurements bromobenzene was found to be the limiting reagent since only 4.56 ml of bromobenzene was used the amount of product predicted to be recovered was 8.7533 g

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in this experiment we will carry out the nitration of bromobenzene bromine substituted on an aromatic ring is an ortho para directing group it will direct an incoming electrophile to the ortho and para positions it is also deactivating meaning that its reactions will be slower than the identical reactions of benzene

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section 20711 lab experiment nitration of bromobenzene introduction the chemical reaction of nitration consists of a nitro group being added to or substituted in a molecule nitration can basically be carried out by a mixture of concentrated nitric acid and sulfuric acid this mixture is useful to obtain the active nitronium ion

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in the nitration of bromobenzene a mixture of sulfuric acid and nitric acid react with one another nitric acid the weaker of the two acids will serve as base to form nitronium ion bromobenzene will attack the nitronium ion by nucleophilic attack

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then actual is your actual findings yield actual theoretical x 100 what are the possible side reactions of nitration of

bromobenzene dinitration can occur at ortho para or at both ortho positions occurs in overheating meta product happens very rarely because bromine is an ortho para director

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the soap in the lab is not suitable for this purpose 2 return to the lab and apply mineral oil to the affected area theory and explanation of the nitration of bromobenzene if the temperature of the reaction mixture exceeds 60 o

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the bromine substituent in bromobenzene is an ortho para director the para position is favored due to steric effects in the reaction that takes place in this lab a nitro group is substituted onto bromobenzene at the para position to form 1 bromo 4 nitrobenzene

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study with quizlet and memorize flashcards containing terms like reacting sulfuric and nitric acid together produces the nitronium ion alone nitric acid is not a strong enough electrophile to react with bromobenzene sulfuric acid is added to the nitric acid the reverse rxn can overheat and boil easily includes a claisen adaptor and

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over 5 minutes slowly pipet this bromobenzene into the icy cold nitration mixture hno 3 h 2 so 4 in the 125 ml flask and stir continue to stir the reaction mixture in the ice bath for 30 minutes do not heat in a new 250 ml erlenmeyer

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on the right the arenium cation has a nitro set with incredibly small 0.6 angstrom bond distances this will result in an optimization failure as the atoms are even close for the programme to accommodate their electrons demonstration of creating the p arenium cation by to nitration of bromobenzene

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the purpose of this lab was to observe where substituents are added on a benzene ring by way of electrophilic aromatic substitution the electrophilic source in the nitration of bromobenzene comes from a nitronium ion NO_2^+ which acts as the

electrophilic species

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nitration of bromobenzene is an electrophilic aromatic substitution reaction the electrophile will be formed from 4 0ml of nitric acid and 4 0ml of sulfuric acid and will then react with 4 5ml of bromobenzene this will produce o bromonitrobenzene and p bromonitrobenzene

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