

Molarity Practice Problems #2

- 1) How many liters of 0.88 M LiF solution can be made with 25.5 grams of solute?

$$M = 0.88 M$$

$$25.5 \text{ g LiF} \times \frac{1 \text{ mol}}{25.94 \text{ g}} = 0.983 \text{ mol}$$

- 2) What is the concentration of a solution that has a volume of 660 mL and contains 33.4 grams of aluminum acetate?

$$L = \frac{0.983 \text{ mol}}{0.88 M} = 1.1 L$$

$$L \cdot M = \frac{\text{mols}}{\text{Liters}} \cdot M$$

$$L = \frac{\text{mol}}{M}$$

Molarity Practice Problems With Answers

Jessica J Manson



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