

12. Molecular Geometry and Bonding Theories

Week 11/12/2021 and Week 12/13/2021, Wednesday, 12/15/2021

How do we determine the shape of a molecule? To determine the geometry of a molecule, we use the following steps:

1. Draw the Lewis structure of the molecule.

1/15/21

Watch the video for Chapter 9 and 10 of your text to understand the steps to determine the geometry of a molecule.

Example 10.1	
<p>Learning Goal: To determine the geometry of a molecule, we use the following steps:</p> <p>1. Draw the Lewis structure of the molecule.</p> <p>2. Determine the number of electron groups around the central atom.</p> <p>3. Determine the number of lone pairs around the central atom.</p> <p>4. Determine the number of electron groups around the central atom.</p> <p>5. Determine the geometry of the molecule.</p>	<p style="text-align: center;">Molecular Geometry</p>
<p>Problem 10.1 Determine the geometry of the molecule.</p>	
<p>Solution: This is the Lewis structure of the molecule.</p>	
<p>Step 1: Draw the Lewis structure of the molecule.</p>	
<p>Step 2: Specify the electron configuration of the central atom.</p> <p style="text-align: center;">(Not yet attempted)</p>	
<p>Step 3: Determine the number of electron groups around the central atom.</p>	
<p>Step 4: Determine the number of lone pairs around the central atom.</p>	
<p>Step 5: Determine the geometry of the molecule.</p>	
<p>Step 6: Determine the geometry of the molecule.</p> <p style="text-align: center;">(Not yet attempted)</p>	

Masteringchemistry Answer Key

Emilie Sanchez



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Table of Contents Masteringchemistry Answer Key

1. Understanding the eBook Masteringchemistry Answer Key
 - The Rise of Digital Reading Masteringchemistry Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Masteringchemistry Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Masteringchemistry Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Masteringchemistry Answer Key
 - Personalized Recommendations
 - Masteringchemistry Answer Key User Reviews and Ratings
 - Masteringchemistry Answer Key and Bestseller Lists

5. Accessing Masteringchemistry Answer Key Free and Paid eBooks
 - Masteringchemistry Answer Key Public Domain eBooks
 - Masteringchemistry Answer Key eBook Subscription Services
 - Masteringchemistry Answer Key Budget-Friendly Options
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 - Masteringchemistry Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Masteringchemistry Answer Key
 - Highlighting and Note-Taking Masteringchemistry Answer Key
 - Interactive Elements Masteringchemistry Answer Key
8. Staying Engaged with Masteringchemistry Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Masteringchemistry Answer Key
9. Balancing eBooks and Physical Books Masteringchemistry Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Masteringchemistry Answer Key
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Masteringchemistry Answer Key
 - Setting Reading Goals Masteringchemistry Answer Key
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Masteringchemistry Answer Key
 - Fact-Checking eBook Content of Masteringchemistry Answer Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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