

NEB- GRADE XII  
Model Question  
2079 (2023)  
Chemistry

*Candidates are required to give their answers in their own words as far as practicable.  
The figures in the margin indicate full marks.*

**Time: 3 hrs.**

**Full Marks : 75**

Attempt all the questions

**Group A**

Rewrite the correct options of each questions in your answer sheet.

**(11x1= 11)**

1. Identify the equivalent weight of  $K_2Cr_2O_7$  in the following reaction?

(Cr =52, K=39)



A) 49

B) 294

C) 98

D) 108

2. In a solution that is at equilibrium, what happens to the concentration of  $H^+$  ions if the concentration of  $OH^-$  ions is increased?

A) The concentration of  $H^+$  ions increases

B) The concentration of  $H^+$  ions decreases

C) The concentration of  $H^+$  ions stay the same

D) It depends on the initial concentration of  $H^+$  ions

3. Assuming the rate of a reaction is doubled for every  $10^\circ C$  rise in temperature how many times increases the rate of temperature rises from  $10^\circ C$  to  $100^\circ C$ ?

A) 112 times

B) 400 times

C) 512 times

D) 614 times

4. for the given reactions



The enthalpy of formation of CO becomes

A)  $2y-x$

B)  $2x-y$

C)  $\frac{y-2x}{2}$

D)  $\frac{x-2y}{2}$

5. What product would be obtained if red hot copper wire reacts with steam?

A)  $CuO$

B)  $Cu_2O$

C)  $Cu_2O_2$

D)  $CuO_2$

6. For which manufacturing process, Bessemer converter is used?

A) Pig iron

B) Steel

C) Wrought iron

D) Cast iron

7. When Sodium phenoxide reacts with methyl bromide it gives

A) Cresol

B) Toluene

C) benzene

D) Anisole

# National Insurance Company Model Question Paper

**Christian Drosten**



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