

Matter Test Study Guide



matter = anything that has mass and takes up space.

pure substance = a single kind of matter that is pure, meaning that it always has a specific chemical makeup. Ex.) table salt is always NaCl and sugar is always $C_{12}H_{22}O_{11}$

element= a pure substance that can't be broken down into any other substance. Ex.) oxygen, copper, aluminum, zinc, tungsten, carbon, hydrogen, helium, nitrogen

atom= the smallest part of an element. Ex.) one atom of oxygen



compound= a pure substance made of two or more elements chemically combined in a set ratio. Ex.) H_2O

molecule= the smallest part of a compound Ex.) one molecule of H_2O



mixture = made of two or more substances that are mixed together, but not chemically combined. Ex.) trail mix

homogeneous= can't see the different parts. Ex.) sugar water

heterogeneous= can see the different parts. Ex.) salad



solution = an example of a homogenous mixture

chemical formulas= tell you which elements and how many atoms of each are in a compound ex.) $C_{12}H_{22}O_{11}$

set ratio = a specific number of atoms of each element to make a compound

weight = a measure of the force of gravity on an object; can change- depends on location
Ex.) earth vs. moon

mass = a measure of the amount of matter in an object; **does not change**

- triple beam balance
- SI unit is the kilogram
- scientists prefer to measure by mass rather than weight because mass doesn't change



My **WEIGHT** on Earth is around 560N



My **WEIGHT** on the moon is around 90N



My **MASS** is always 56kg!

Matter Study Guide

Kenneth W. Raymond



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Chemistry: Matter & Change, Study Guide For Content Mastery, Student Edition McGraw Hill,2001-03-30 Study Guide and Reinforcement Worksheets allow for differentiated instruction through a wide range of question formats There are worksheets and study tools for each section of the text that help teachers track students progress toward understanding concepts Guided Reading Activities help students identify and comprehend the important information in each chapter

Glencoe Chemistry Glencoe/McGraw-Hill,2002 **Holt Chemistry** ,2000 **Resources in Education** ,1984

Chemistry, Study Guide Neil D. Jespersen,James E. Brady,Alison Hyslop,2011-05-03 Chemistry The Study of Matter and Its Changes Sixth Edition will provide the necessary practice support and individualised instruction that ensures success in the General Chemistry course This text provides the forum for problem solving and concept mastery of chemical phenomena that leads to proficiency and success in the General Chemistry course This edition will continue a molecular basis of chemistry tradition but in a manner that overtly and repeatedly reinforces the way properties at the molecular level are related to properties we observe at the macroscopic level The unique chemical tools approach employed in this book provides a way of thinking that helps those students develop the ability to analyse and solve both mathematical and conceptual problems This text follows the successful three step approach described as ANALYSIS SOLUTION and IS THE ANSWER REASONABLE This encourages the student to think about the problem before attempting to solve it then working through the solution and finally asking the important question Does the answer make sense There are problem sets called Bringing It Together that contain problems which require students to bring together concepts from two or more of the preceding chapters This reinforces learned concepts and builds concept mastery Chemistry, Study Guide James E. Brady,John R. Holum,1996-03-01 Offers accurate lucid and interesting explanations of basic concepts and facts of chemistry while helping readers develop skills in analytical thinking and problems solving **Study Guide and Partial Solutions Manual, Fundamentals of General, Organic, and Biological Chemistry** Susan McMurry,1996 This internationally acclaimed detective series is just the thing for lovers of thoseNumber One Ladieslooking for a darker more realistic view of Botswana Sue Baker Publishing News **Student Study Guide and Solutions Manual to accompany General Organic and Biological Chemistry, 1e** Kenneth W. Raymond,2005-10-07 Finally readers have a shorter less intimidating introduction to general organic and biological chemistry Not only is Raymond s text concise it also takes an integrated approach to presenting important topics in a way that makes the material easier to understand In this approach similarities can be exploited and concepts reinforced The result is that readers see the strong connections that exist between these three branches of chemistry **Matter and Energy** ,2005 Review Notes and Study Guide to Hemingway's Farewell to Arms, and The Sun Also Rises, For Whom the Bell Tolls [and] The Old Man and the Sea Stanley Cooperman,1964 Professional Cooking , Study Guide Wayne Gisslen,1998-09-07 Completely redesigned updated and richly illustrated revision of the classic

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