



Name: _____

Date: _____

Student Exploration: Half-life

Background: Scientists study atoms, which are super-tiny particles. Half-life is another word for radioactive decay.

Prerequisites: **Scientific Method** and **Understanding the Basics**

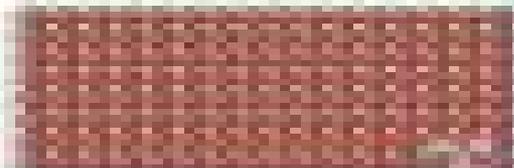
1. What do you think these numbers represent? How, exactly, do they define the property of the substance? 1000 particles, 100%

2. If you have the radioactive substance, is the rate of popping through the barrier or how it changes? It changes. It changes from 100% to 50% and 25% and 12.5% and 6.25%.

Gizmo Warm-up

The simulation simulates the process of a substance with radioactive atoms. Radioactive atoms release energy through particles. In this simulation, you perform virtual experiments with isotopes that release alpha particles.

The half-life of a substance is the time it takes for the number of radioactive particles to decrease to half of its original amount. In this simulation, you can:



Remaining **Released**

3. What happens over time? 1000 particles, goes down to 500 particles, 250 particles.

How does the number around you decrease from a single particle, an enormous substance, the particles and energy emitting (alpha) particles.

4. What happens at the end of the decay process? Nothing more.

5. Is the rate of decay constant the throughout, or does it change? It changes.



Explore Learning Gizmo Answer Key Half Life

Raffaela Di Napoli



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Adopting the Beat of Expression: An Emotional Symphony within **Explore Learning Gizmo Answer Key Half Life**

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