

Grade 7

Natural Sciences

Term 4

Exam paper 1

Section	Questions	Answers
Section A Name 3 elements: _____ Instructions: 1. Label each circle with an element name. 2. Write an atomic symbol under each name. 3. Write the mass number and atomic number of each element. 4. Write the number of protons and neutrons in each nucleus. 5. Write the number of electrons in each atom. 	QUESTION 1 Diagram 1 shows a neutral atom with 11 protons and 11 electrons. The atom is represented by a central nucleus and three concentric shells. 1.1 What is the best description for the atom? 1.2 How many outer shell electrons does the atom have? 1.3 What is the atomic number of the atom? 1.4 Why do atoms usually combine to form molecules? 1.5 What is the mass number of the atom? 1.6 How many protons and neutrons are in the nucleus? 1.7 How many electrons are in the atom? 1.8 What is the name of the atom? 1.9 How many shells does the atom have? 1.10 How many protons and neutrons are in the nucleus? 1.11 How many electrons are in the atom? 1.12 How many shells does the atom have? 1.13 How many protons and neutrons are in the nucleus? 1.14 How many electrons are in the atom? 1.15 How many shells does the atom have? 1.16 How many protons and neutrons are in the nucleus? 1.17 How many electrons are in the atom? 1.18 How many shells does the atom have? 1.19 How many protons and neutrons are in the nucleus? 1.20 How many electrons are in the atom?	ANSWERS 1.1 A neutral atom 1.2 1 1.3 11 1.4 To achieve a stable electron configuration 1.5 23 1.6 11 protons and 12 neutrons 1.7 11 1.8 Sodium 1.9 3 1.10 11 protons and 12 neutrons 1.11 11 1.12 3 1.13 11 protons and 12 neutrons 1.14 11 1.15 3 1.16 11 protons and 12 neutrons 1.17 11 1.18 Sodium 1.19 3 1.20 11 protons and 12 neutrons



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