

## EQUILIBRIUM CONSTANT - WORKSHEET 2\_ANSWERS

1. Write the equilibrium constant expression for each of the following reactions:



$$K = \frac{[NO_2]^2}{[N_2][O_2]}$$



$$K = \frac{[C_2H_6]}{[H_2]^3}$$



$$K = \frac{1}{[I_2(aq)]}$$

2. For the following reaction:



At equilibrium,  $[Fe^{2+}] = 0.50 M$ ,  $[Ag^+] = 1.0 M$ ,  $[Fe^{3+}] = 1.50 M$

a) Calculate the equilibrium constant for the reaction.

$$K = 3.0$$

b) If at equilibrium,  $[Fe^{2+}] = 0.20 M$  and  $[Ag^+] = 0.30 M$ , what must be the concentration of  $Fe^{3+}$ ?

$$[Fe^{3+}] = 0.18 \text{ mol/L}$$



Calculate the equilibrium concentration of oxygen gas if the NO and  $NO_2$  concentrations are equal at equilibrium.

$$[O_2] = 2.0 \times 10^{-6} \text{ mol/L}$$



Which of the above reactions would have the greatest amount of products present at equilibrium compared to reactants?

Reaction b. Since K is larger the amount of products must be larger, because K is products/reactants.

5. For the reaction:



at an equilibrium temperature of  $1000^\circ C$ , a 1.00 L flask contains 0.120 moles of ammonia, 1.03 moles of nitrogen and 1.62 moles of hydrogen. Calculate the equilibrium constant at this temperature.

$$K = 3.29 \times 10^{-2}$$

# Genetic Equilibrium Worksheet Answer Key

**Scott C. Dulebohn**



**Genetic Equilibrium Worksheet Answer Key:**

## Unveiling the Power of Verbal Artistry: An Mental Sojourn through **Genetic Equilibrium Worksheet Answer Key**

In a global inundated with monitors and the cacophony of quick conversation, the profound energy and mental resonance of verbal beauty frequently fade into obscurity, eclipsed by the continuous assault of sound and distractions. However, nestled within the musical pages of **Genetic Equilibrium Worksheet Answer Key**, a charming work of literary beauty that impulses with raw thoughts, lies an wonderful journey waiting to be embarked upon. Published by way of a virtuoso wordsmith, this mesmerizing opus manuals viewers on a mental odyssey, gently revealing the latent possible and profound impact embedded within the elaborate web of language. Within the heart-wrenching expanse with this evocative analysis, we shall embark upon an introspective exploration of the book is main styles, dissect its captivating writing model, and immerse ourselves in the indelible impression it leaves upon the depths of readers souls.

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