

GRAPHING TRENDS OF THE PERIODIC TABLE

CHEMISTRY

NAME: _____

DATE: _____

PERIOD: _____ Graphing Periodic Table Trends

Directions: Using the provided graph paper, graph each of the elemental properties using a line graph. Place the property on the Y-axis and the Atomic number on the X-axis. Draw a dark, solid, vertical line on each graph where each row on the periodic table ends. (i.e. after atomic number 20). Use the graphs to analyze the trends in these properties and to answer the questions on the back.

Atomic Radius (pm)							
Lithium - 3	157	Sodium - 11	190	Potassium - 19	243	Rubidium - 27	295
Beryllium - 4	112	Magnesium - 12	145	Calcium - 20	194	Strontium - 28	219
Boron - 5	87	Aluminum - 13	175	Gallium - 31	136	Indium - 40	156
Carbon - 6	67	Silicon - 14	111	Germanium - 32	125	Tin - 50	145
Nitrogen - 7	56	Phosphorus - 15	98	Arsenic - 33	114	Antimony - 51	133
Oxygen - 8	48	Sulfur - 16	88	Selenium - 34	103	Tellurium - 52	123
Fluorine - 9	42	Chlorine - 17	79	Bromine - 35	94	Iodine - 53	119

Ionization Energy (kJ/mol)							
Lithium - 3	520	Sodium - 11	496	Potassium - 19	419	Rubidium - 27	403
Beryllium - 4	900	Magnesium - 12	738	Calcium - 20	590	Strontium - 28	550

Atomic Radius



Investigating And Graphing Periodic Trends Answer Key

**Zvi Szafran, Ronald M. Pike, Judith C.
Foster**

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Table of Contents Investigating And Graphing Periodic Trends Answer Key

1. Understanding the eBook Investigating And Graphing Periodic Trends Answer Key
 - The Rise of Digital Reading Investigating And Graphing Periodic Trends Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Investigating And Graphing Periodic Trends Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Investigating And Graphing Periodic Trends Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Investigating And Graphing Periodic Trends Answer Key
 - Personalized Recommendations
 - Investigating And Graphing Periodic Trends Answer Key User Reviews and Ratings
 - Investigating And Graphing Periodic Trends Answer Key and Bestseller Lists
5. Accessing Investigating And Graphing Periodic Trends Answer Key Free and Paid eBooks
 - Investigating And Graphing Periodic Trends Answer Key Public Domain eBooks
 - Investigating And Graphing Periodic Trends Answer Key eBook Subscription Services
 - Investigating And Graphing Periodic Trends Answer Key Budget-Friendly Options

6. Navigating Investigating And Graphing Periodic Trends Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Investigating And Graphing Periodic Trends Answer Key Compatibility with Devices
 - Investigating And Graphing Periodic Trends Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Investigating And Graphing Periodic Trends Answer Key
 - Highlighting and Note-Taking Investigating And Graphing Periodic Trends Answer Key
 - Interactive Elements Investigating And Graphing Periodic Trends Answer Key
8. Staying Engaged with Investigating And Graphing Periodic Trends Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Investigating And Graphing Periodic Trends Answer Key
9. Balancing eBooks and Physical Books Investigating And Graphing Periodic Trends Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Investigating And Graphing Periodic Trends Answer Key
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Investigating And Graphing Periodic Trends Answer Key
 - Setting Reading Goals Investigating And Graphing Periodic Trends Answer Key
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Investigating And Graphing Periodic Trends Answer Key
 - Fact-Checking eBook Content of Investigating And Graphing Periodic Trends Answer Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

-
- Interactive and Gamified eBooks

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