

Binary Ionic Compounds

Name: _____, Per: _____

Directions: First quickly scan the worksheet and write any metals you are confused on as a metal or as a nonmetal that is a transition metal. Then either give the name or chemical formula as necessary for each problem below.

Remember: that transition metals can have multiple oxidation states, so you are required to include the appropriate oxidation state in parentheses when naming them. This does not use the suffixes and oxidation state of the anion to help determine the oxidation state of the transition metal.

- | | |
|-----------------------------------|-----------------------------------|
| 1. LiF _____ | 2. lithium chloride _____ |
| 3. LiOH _____ | 4. lithium oxide _____ |
| 5. Li_2O _____ | 6. magnesium fluoride _____ |
| 7. NaCl _____ | 8. magnesium sulfate _____ |
| 9. BaF_2 _____ | 10. iron (II) chloride _____ |
| 11. BaSO_4 _____ | 12. iron (III) _____ |
| 13. BaO _____ | 14. sodium fluoride _____ |
| 15. Ca_3N_2 _____ | 16. copper (II) chloride _____ |
| 17. Ca_3N_2 _____ | 18. copper (II) oxide _____ |
| 19. Ca_3N_2 _____ | 20. copper (II) chloride _____ |
| 21. Na_2O _____ | 22. sodium chloride _____ |
| 23. PbO_2 _____ | 24. lead (IV) fluoride _____ |
| 25. PbO _____ | 26. lead (IV) sulfate _____ |
| 27. Pb_2O_3 _____ | 28. lead (IV) oxide _____ |
| 29. Fe_2O_3 _____ | 30. iron (II) fluoride _____ |
| 31. Fe_2O_3 _____ | 32. aluminum (IV) phosphate _____ |
| 33. Fe_2O_3 _____ | 34. aluminum fluoride _____ |
| 35. Al_2O_3 _____ | 36. aluminum oxide _____ |
| 37. MnO_2 _____ | 38. manganese (IV) fluoride _____ |
| 39. SnO_2 _____ | 40. tin (IV) oxide _____ |

Liz Larosa Science Scrambles

K Morrison



Liz Larosa Science Scrambles:

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