

# Midsegment Theorem and Triangle Inequality

Find the missing length indicated.

1) Find  $RP$



2) Find  $MK$



3) Find  $XV$



4) Find  $YW$

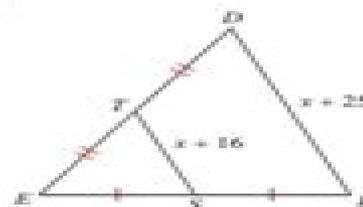


Solve for  $x$ .

5)



6)



# Midsegment Worksheet Answer

**Victor M. Corman**



**Midsegment Worksheet Answer:**

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the objectives of the boiling point determination lab were to determine the boiling point of a mystery substance provided to each student boiling point is the point at which a liquid changes physical states and becomes a vapor or gas boiling point is important to understand because it can be used to help distinguish between different substances

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the boiling point of an organic liquid is the temperature at which its vapour pressure equals the atmospheric pressure over the liquid or it is the temperature at which the vapour and liquid phases are in equilibrium at agiven pressure the boiling point is considered as a criterion of purity of a compound and is useful for identification of

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given that the boiling points of ethanol and methanol 78 c and 64 c respectively are far off in a significant way this seems like

a valid conclusion thermometer calibrations are necessary in order to collect reliable information as precise calculations are imperative to gathering correct results

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melting point 1 a push the open end of a capillary tube into the powdered acetamide b move the powder to the closed end of the capillary tube by tapping it on the table repeat until the the powdered acetamide occupies 1 2 mm of the capillary tube end c

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boiling point determination lab report conclusion objective to correctly measure boiling issues of various liquids and identify an unknown liquid based on its boiling point background information objekte can be distinguished into acid properties and physical properties

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discussion and conclusion the technique recrystallization was used because it is a simple method to purify a solid compound the percentage yield for benzoic acid and acetanilide acid being high indicated the initial sample had a low concentration of impurities with the melting point being in a specific

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introduction in this lab we will be using boiling points melting points and infrared spectrometry to identify unknown compounds boiling points are affect by three key factors the first being intermolecular forces inter molecular affect boiling point based on bond strength

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determination of boiling points introduction the boiling point of a compound is the temperature at which it changes from a liquid to a gas this is a physical property often used to identify substances or to check the purity of the compound it

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clean up solid and liquid spills immediately true g conclusions after conducting the boiling point determination lab it can be concluded that the unknown substance 193 is isopropyl alcohol on account of its observed boiling point of 88 c

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conclusion after completing this experiment the melting point of three different unknown compounds was obtained using the mel temp apparatus compound a had a melting point range of 114 6 115 5 c thus revealing it to be acetanilide compound b had a melting point range of 122 6 124 5 c thus

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nov 27 2019 essay sample the purpose of this experiment is to determine the boiling points of various organic compounds

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melting point can be used as the identity and purity of an organic solid however it is not the same as the boiling point the boiling point is a characteristic unique of pure liquids the boiling point is the temperature at which the vapor pressure of a liquid is equal to the atmospheric pressure

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aug 11 2020 boiling points are very sensitive to changes in applied pressure so all boiling points should be reported with the measured pressure a compound's normal boiling point refers to its boiling point at a pressure of 760 mm Hg

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the purpose of the experiment is to understand what vapour pressure is and how it changes with temperature to understand the relationship between vapour pressure at a given temperature and boiling temperature at a given pressure to determine the vapour pressure of a pure liquid at various temperatures

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exploring boiling points abstract in this experiment we are given a set of straight chain alkanes and a set of compounds to compare by following the trendline of the straight chain alkanes it could be concluded that as the molar mass of the straight chain alkanes increases their boiling points increase as well

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dec 7 2019 1987 the sample essay on boiling point determination lab report deals with a framework of research based facts approaches and arguments concerning this theme to see the essay's introduction body paragraphs and conclusion read on should be based on observations made during the experiment

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laboratory partners samantha and mackenzie goals □ correctly determine the boiling point of an unknown liquid □ measure the refractive index of an unknown □ use the gas chromatogram to determine the mass percentage of a mixture and

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50 °C so in order to convert an experimental measurement taken in Calgary which is at higher altitude than sea level and so is at a lower pressure to that reported for sea level higher pressure one needs to add a correction factor since the boiling point at sea level is higher than that at higher altitudes

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introduction in chemistry one may encounter an unknown substance that needs to be determined by physical properties such as boiling point melting point density solubility etc in this experiment I am to figure out the boiling point of a mystery

substance in order to identify its chemical responsibility this experiment requires the handling of a bunsen burner boiling  
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there are a variety of methods by which a sample's boiling point can be determined including distillation reflux and by using a thiele tube the most straightforward method uses a thiele tube and has the advantage of using less than 0.5 ml of material

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web easy serves 4 make it dairy free use vegan cream cheese for cream cheese unsweetened coconut cream for heavy cream and unsweetened oat milk for whole milk whisk the coconut cream until smooth then add the remaining ingredients make it lite use  $\frac{1}{2}$  teaspoon stevia and  $2\frac{1}{2}$  tablespoons raw agave nectar for granulated sugar

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web jan 2 2023 the ninja creami recipe makes one pint sized jar of ice cream but you can also purchase multiple pints to create several containers of vanilla ice cream at once this is great for having that base flavor and then using mix

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web may 16 2023 this ninja creami vanilla ice cream is kind of a mix between a no churn recipe and a traditional ice cream recipe but without the hassle of cooking or churning the ice cream all the ingredients are added to a blender and then poured into a pint size container to freeze for 24 hours

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web may 28 2023 ingredients fairlife core power vanilla protein shake this is the 14 fl oz vanilla shake core power is made with ultra filtered gluten free high protein lactose free milk stevia sugar substitute sugar free instant vanilla pudding mix this is a dry mix that can be found in grocery stores on the baking aisle

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web step one prepare your ice cream mixture add the cream cheese sugar and vanilla bean paste in a large mixing bowl mix together until the granulated sugar is dissolved fold in the heavy whipping cream and milk and mix well until all ingredients

are combined

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web nov 24 2022 this recipe is spectacular it comes out creamier and richer than the standard vanilla recipe from the ninja creami recipe booklet using real dairy just made it for my husband who is very picky and he loved it thought it tasted very close to regular vanilla ice cream

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web may 29 2023 ingredients low fat milk you can use nonfat milk but know the flavor and texture will be slightly impacted you can also use whole milk my favorite milk to use is the fairlife reduced fat milk that is lactose free instant vanilla pudding mix you can use regular or sugar free there are two reasons for adding a little bit of this

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web may 22 2022 ingredients 1 cup whole milk 1 cup heavy cream 1 4 cup of sweetened condensed milk 1 tbsp of pure liquid vanilla extract 1 4 cup of granulated sugar 1 4 cup of powdered sugar 1 pinch of salt

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web this usually does the trick if you need to make this dairy free you can use unsweetened oat milk instead of milk unsweetened coconut cream instead of heavy cream and vegan cream cheese instead of regular cream cheese recipe slightly adapted from ninja creami

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web jul 27 2023 for the full vanilla ice cream recipe for the ninja creami with measurements see the recipe card at the end of the post place your half and half sugar vanilla extract and a pinch of salt into a blender

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web in a large microwave safe bowl microwave the cream cheese for 10 seconds add the sugar and vanilla and combine until the mixture is smooth and creamy slowly mix in the heavy cream and milk until fully combined and sugar is dissolved pour base into an empty creami pint

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web in a small mixing bowl or large liquid measuring cup combine instant pudding mix and milk whisk together until pudding mix dissolves add heavy cream and vanilla extract whisking just to combine transfer mixture to a ninja creami pint being careful not to exceed the max fill line and close the lid

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