

General Chemistry Semester 2 Study Guide

Mr. Price - Fall 2016

1st Period: Monday, June 27th, 8:00am - 10:00am

2nd Period: Wednesday, June 29th, 11:00am - 1:00pm

3rd Period: Thursday, June 30th, 11:00am - 1:00pm

Use your resources wisely throughout the semester. You may not get to the end of the book because of a variety of circumstances that will be your task throughout the semester. The following notes and resources are for you to use when you are unable to go to class. Use a variety of resources, but I would suggest the following: Khan Academy, YouTube, and Google for additional notes.

Make yourself a reading schedule of the material covered throughout the semester. This is just a guide and I do suggest the course schedule I have made that will make your studying more efficient. Use the completed schedule to identify whether you have read what you need additional practice.

Be able to: describe, explain, and predict chemical reactions, stoichiometry, equilibrium, and thermodynamics.

Study your lab notes.

Make sure you have the following resources necessary available and that you have access to your assignments throughout the semester.

- 1. Periodic Table
- 2. Chemical Properties Table
- 3. Molar Masses Table
- 4. Thermodynamic Tables
- 5. Naming and Formula of Compounds Table (Cation, Anion, Binary, Ternary, Polyatomic)
- 6. Solubility Rules Table (necessary for reactions)
 - a. aq and aq
 - b. aq and aq
 - c. aq and aq
- 7. Chemical Equations (Balanced Equations)

Excellent Reading Material

Below are some links for additional resources and information about chemistry. Make sure to check all appropriate resources.

- 1. Molecular Structure of Gases (10/1/16)
 - a. Understanding Avogadro's Law
 - b. Molecular Weight
 - c. Molecular Weight
- 2. Molecular Weight (10/1/16) and (10/1/16)
 - a. Molecular Weight
 - b. Molecular Weight
- 3. Thermodynamic Tables (10/1/16)
 - a. Thermodynamic Tables
 - b. Specific Heat
- 4. Equilibrium (10/1/16)
 - a. Equilibrium (10/1/16)
 - b. Equilibrium (10/1/16)
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