

# GRAPHING TRENDS OF THE PERIODIC TABLE

## CHEMISTRY

NAME: \_\_\_\_\_

DATE: \_\_\_\_\_

### PERIOD: \_\_\_\_\_ Graphing Periodic Table Trends

**Directions:** Using the provided graph paper, graph each of the elemental properties using a line graph. Place the property on the Y-axis and the Atomic number on the X-axis. Draw a dark, solid, vertical line on each graph where each row on the periodic table ends. (i.e. after atomic number 20). Use the graphs to analyze the trends in these properties and to answer the questions on the back.

Atomic Radius (pm)							
Lithium - 3	157	Sodium - 11	190	Potassium - 19	243	Rubidium - 27	295
Beryllium - 4	112	Magnesium - 12	145	Calcium - 20	194	Strontium - 28	219
Boron - 5	87	Aluminum - 13	118	Gallium - 31	136	Indium - 49	156
Carbon - 6	67	Silicon - 14	111	Germanium - 32	125	Tin - 50	145
Nitrogen - 7	56	Phosphorus - 15	98	Arsenic - 33	114	Antimony - 51	133
Oxygen - 8	48	Sulfur - 16	88	Selenium - 34	103	Tellurium - 52	123
Fluorine - 9	42	Chlorine - 17	79	Bromine - 35	94	Iodine - 53	119

Ionization Energy (kJ/mol)							
Lithium - 3	520	Sodium - 11	496	Potassium - 19	419	Rubidium - 27	403
Beryllium - 4	900	Magnesium - 12	738	Calcium - 20	590	Strontium - 28	550

### **Atomic Radius**



# Graphing Trends In The Periodic Table Key

**Michael J. Padilla**



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Chemistry in the Community (Enhanced Core Four) American Chemical Society,2006-02-15 *Chemistry Homework for OCR A for Double and Separate Awards* Gareth Pritchard,2001 Howework activites for OCR A Chemistry specifications

**ChemDiscovery Student Guide** Olga I. Agapova,2002 **Prentice Hall Science Explorer** Michael J. Padilla,2002

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by depicting it as a seamless discipline the text shows how organic inorganic and physical concepts can be blended together in order to achieve the common goal of understanding chemical systems With a lively and engaging writing style enhanced by vivid illustrations only Chemical Structure and Reactivity makes teaching chemistry with an integrated approach possible Special Features The only introductory text to take a truly integrated approach in explaining the fundamentals of chemistry Fosters an orbital based understanding of reactions with clear curly arrow mechanistic detail throughout A two part structure allows flexibility of use Part I lays down the core of the subject while Part II describes a series of relatively standalone topics which can be selected to fit a particular course Numerous concepts are illustrated with fully cross referenced custom developed online modules enabling students to develop an understanding through active learning Self test exercises embedded in the text with solutions at the end of each chapter and extensive question sets encourage hands on learning to help students master the subject and gain confidence The Online Resource Centre features a range of additional resources for both students and registered adopters of the book New to this Edition A new chapter on symmetry has been added to Part I Discussions of organometallic chemistry spectroscopy and molecular geometry have been expanded Cross references from Part I to Part II have been increased to make the links between core concepts and more advanced topics clearer More self test questions and exercises have been provided

Modern Physics Gary N. Felder, Kenny M. Felder, 2022-09-15 Modern Physics intertwines active learning pedagogy with the material typically covered in an introductory survey from the basics of relativity and quantum mechanics through recent developments in particle physics and cosmology The flexible approach taken by the authors allows instructors to easily incorporate as much or as little active learning into their teaching as they choose Chapters are enhanced by Discovery and Active Reading exercises to guide students through key ideas before or during class while ConcepTests help check student understanding and stimulate classroom discussions Each chapter also includes extensive assessment material with a range of basic comprehension questions drill and practice calculations computer based problems and explorations of advanced applications A test bank and interactive animations as well as other support for instructors and students are available online Students are engaged by an accessible and lively writing style thorough explanations Math Interludes which account for varying levels of skill and experience and advanced topics to further pique their interest in physics

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 Ralph Thomas Myers, 2004 *Topics-aids* Rod O'Connor, 1975 Solutions Manual to Accompany General Chemistry, Fourth Edition, and General Chemistry with Qualitative Analysis, Fourth Edition, Whitten, Gailey, Davis Yi-Noo Tang, Wendy L. Keeney-Kennicutt, 1992 *Australian Journal of Chemistry*, 2004 **Reviews in Graph Theory** William G. Brown, 1980  
*Proceedings of the International Conference on Fast Breeder Systems*, 1987 **Chemistry** Martin Stuart  
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