

GRAPHING TRENDS OF THE PERIODIC TABLE

CHEMISTRY

NAME: _____

DATE: _____

PERIOD: _____ Graphing Periodic Table Trends

Directions: Using the provided graph paper, graph each of the elemental properties using a line graph. Place the property on the Y-axis and the Atomic number on the X-axis. Draw a dark, solid, vertical line on each graph where each row on the periodic table ends. (i.e. after atomic number 20). Use the graphs to analyze the trends in these properties and to answer the questions on the back.

| Atomic Radius (pm) | | | | | | | |
|--------------------|-----|-----------------|-----|----------------|-----|----------------|-----|
| Lithium - 3 | 152 | Sodium - 11 | 190 | Potassium - 19 | 243 | Rubidium - 27 | 295 |
| Beryllium - 4 | 112 | Magnesium - 12 | 145 | Calcium - 20 | 194 | Strontium - 28 | 219 |
| Boron - 5 | 87 | Aluminum - 13 | 175 | Gallium - 21 | 136 | Indium - 49 | 156 |
| Carbon - 6 | 67 | Silicon - 14 | 111 | Germanium - 32 | 125 | Tin - 50 | 145 |
| Nitrogen - 7 | 56 | Phosphorus - 15 | 98 | Arsenic - 33 | 114 | Antimony - 51 | 133 |
| Oxygen - 8 | 48 | Sulfur - 16 | 88 | Selenium - 34 | 103 | Tellurium - 52 | 123 |
| Fluorine - 9 | 42 | Chlorine - 17 | 79 | Bromine - 35 | 94 | Iodine - 53 | 119 |

| Ionization Energy (kJ/mol) | | | | | | | |
|----------------------------|-----|----------------|-----|----------------|-----|----------------|-----|
| Lithium - 3 | 520 | Sodium - 11 | 496 | Potassium - 19 | 419 | Rubidium - 27 | 403 |
| Beryllium - 4 | 900 | Magnesium - 12 | 738 | Calcium - 20 | 590 | Strontium - 28 | 550 |

Atomic Radius



Graphing Periodic Trends Lab Answer Key

Marco Cascella



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Laboratory Manual Jo A. Beran,1990 **Computer Based Projects for a Chemistry Curriculum** Thomas J. Manning,Aurora P. Gramatges,2013-04-04 This e book is a collection of exercises designed for students studying chemistry courses at a high school or undergraduate level The e book contains 24 chapters each containing various activities employing applications such as MS excel spreadsheets and Spartan computational modeling Each project is explained in a simple easy to understand manner The content within this book is suitable as a guide for both teachers and students and each chapter is supplemented with practice guidelines and exercises Computer Based Projects for a Chemistry Curriculum therefore serves to bring computer based learning a much needed addition in line with modern educational trends to the chemistry classroom

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