

GRAPHING TRENDS OF THE PERIODIC TABLE

CHEMISTRY

NAME: _____

DATE: _____

PERIOD: _____ *Graphing Periodic Table Trends*

Directions: Using the provided graph paper, graph each of the elemental properties using a line graph. Place the property on the Y-axis and the Atomic number on the X-axis. Draw a dark, solid, vertical line on each graph where each row on the periodic table ends. (i.e. after atomic number 20). Use the graphs to analyze the trends in these properties and to answer the questions on the back.

| Atomic Radius (pm) | | | | | | | |
|--------------------|-----|-----------------|-----|----------------|-----|----------------|-----|
| Lithium - 3 | 157 | Sodium - 11 | 190 | Potassium - 19 | 243 | Rubidium - 27 | 295 |
| Beryllium - 4 | 112 | Magnesium - 12 | 145 | Calcium - 20 | 194 | Strontium - 28 | 219 |
| Boron - 5 | 87 | Aluminum - 13 | 118 | Gallium - 31 | 135 | Indium - 40 | 156 |
| Carbon - 6 | 67 | Silicon - 14 | 111 | Germanium - 32 | 125 | Tin - 50 | 145 |
| Nitrogen - 7 | 56 | Phosphorus - 15 | 98 | Arsenic - 33 | 114 | Antimony - 51 | 133 |
| Oxygen - 8 | 48 | Sulfur - 16 | 88 | Selenium - 34 | 103 | Tellurium - 52 | 123 |
| Fluorine - 9 | 42 | Chlorine - 17 | 79 | Bromine - 35 | 94 | Iodine - 53 | 119 |

| Ionization Energy (kJ/mol) | | | | | | | |
|----------------------------|-----|----------------|-----|----------------|-----|----------------|-----|
| Lithium - 3 | 520 | Sodium - 11 | 496 | Potassium - 19 | 419 | Rubidium - 27 | 403 |
| Beryllium - 4 | 900 | Magnesium - 12 | 738 | Calcium - 20 | 590 | Strontium - 28 | 550 |

Atomic Radius



Graphing Periodic Trends Answer Key

Ralph Thomas Myers



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Academic Computing ,1987

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